

Spin-Canting Magnetization in an Unusual Co₄ Cluster-Based Layer Compound from a 2,3-Dihydroxyquinoxaline LigandChen-I Yang,[†] Po-Hsiang Chuang,^{†,‡} Gene-Hsiang Lee,[§] Shie-Ming Peng,[§] and Kuang-Lieh Lu^{*,†}[†]Institute of Chemistry, Academia Sinica, Taipei 115, Taiwan[‡]Graduate Institute of Engineering, National Taiwan University of Science and Technology, Taipei 106, Taiwan[§]Department of Chemistry, National Taiwan University, Taipei 106, Taiwan

Supporting Information

ABSTRACT: The self-assembly of Co(O₂CPh)₂ with a 2,3-dihydroxyquinoxaline (H₂dhq) linker has revealed a new two-dimensional cluster-based compound, [Co₄(OMe)₂(O₂CPh)₂(dhq)₂(MeOH)₂]_n, which shows spin-canted magnetization and a definite magnetic hysteresis loop.

One of the major challenges in the area of molecular magnetism is the construction of magnetic frameworks that show spontaneous magnetization. In studies of the magnetic properties of magnetic frameworks, the main advantage is the availability of numerous types of structures and topologies, in which a large magnetic anisotropy is produced from a single ion and/or there is a situation where structural distortion can result in the production of multifunctional materials with magnetic properties. Several different strategies have been employed, in attempts to achieve such magnetic entities.^{1,2} However, the design and synthesis of such molecular materials with predictable magnetic properties continues to remain a challenging task because structural factors governing exchange coupling between paramagnetic centers are complex and elusive.

Despite the reports of some molecule-based magnetic metal–organic frameworks (MOFs) with a spontaneous magnetization, explorations that are relevant to achieving cluster-based magnetic MOFs are relatively rare in the literature.³ Concerning the design of hybrid magnetic materials, the selection of an appropriate bridging ligand is of primary importance; the ligands may function as superexchange pathways and building units of structural topology. For a ligand such as 2,3-dihydroxyquinoxaline, the two hydroxyl groups and two nitrogen-donor atoms have the capacity to not only connect the metal ions to a multidimensional structure but also shorten the distance between metal ions, which could dominate a magnetic exchange pathway to a significant extent.

We herein report on the production of a new cluster-based two-dimensional (2D) layer, [Co₄(OMe)₂(O₂CPh)₂(dhq)₂(MeOH)₂]_n (**1**), produced from the hydrothermal reaction of 2,3-dihydroxyquinoxaline and Co(O₂CPh)₂. This compound not only shows a clustered type of layer structure from the unusual Co₄ building units but also exhibits spontaneous magnetization with a definite coercivity.

The solvothermal treatment of a mixture of H₂dhq and Co(O₂CPh)₂ in a molar ratio of 1:1 in MeOH at 160 °C in a

Teflon-lined stainless steel autoclave for 3 days resulted in the formation of very fine purple crystals of **1** in a yield of 21%.

An X-ray crystal structural analysis⁴ showed that **1** crystallizes in the monoclinic space group *P2₁/c*. Compound **1** adopts a 2D layer structure that is comprised of a tetranuclear Co^{II} cluster repeating unit (see Figure 1, top). In each unit, four cobalt ions are located in the same plane, with a zigzag arrangement in which the angle for Co1–Co2–Co2' is 105.84(2)°. Co1 and Co2 are coordinated in the CoN₂O₂ tetrahedral and CoNO₅ octahedral geometries, respectively. Peripheral ligation around the Co₄ unit is provided by two PhCO₂[−], two MeO[−], two dhq^{2−} ligands, and two MeOH molecules. These two dhq^{2−} ligands function as a η¹:η¹:η²:η¹:μ₄ binding mode, which chelates Co1 and Co2' terminally with one nitrogen atom and one hydroxyl group, respectively, while Co2 and Co2' are bridged through the other hydroxyl group, resulting in a planar [Co₂(OR)₂] diamond core with a Co–Co distance of 3.252(1) Å. The MeO[−] and benzoate groups show η²:μ₂ and syn,syn η¹:η¹:μ₂ binding modes, respectively, and bridge Co1 and Co2, resulting in a Co1–Co2 distance of 3.082(1) Å. Moreover, the nitrogen atom in dhq^{2−} is further connected to Co1' in the next repeating Co₄ unit to form a 2D layer, in which each alternative neighboring unit has an angle of 68° in the Co₄ plane of the structure (see Figure 1, bottom). In the crystal, the layers are well isolated, with the shortest interlayer Co...Co distance being 9.879 Å, and no significant interlayer interactions are observed (Figure S1 in the Supporting Information).

Variable-temperature direct-current (dc) magnetic susceptibility data were collected for complex **1** in the temperature range of 1.8–300 K under an applied field of 1 kOe, and the data (see Figure 2) were plotted as χ_MT vs *T*. The room-temperature value of χ_MT was 12.2 emu K mol^{−1}, and χ_MT decreases with decreasing temperature to a minimum value of 4.1 emu K mol^{−1} at 20 K, after which it increases, reaching a maximum value of 36.9 emu K mol^{−1} at 10 K, and then decreases again to 8.4 emu K mol^{−1} at 1.8 K. The initial χ_MT could be a result of the depopulation of higher Kramers doublets from the Co^{II} center with an octahedral coordination environment,⁵ and antiferromagnetic coupling within the Co₄ unit. The steep increases in χ_M and χ_MT below 20 K are clearly indicative of spontaneous magnetization, indicating spin-canted antiferromagnetism, which would lead to weak ferromagnetism

Received: July 29, 2011

Published: December 22, 2011

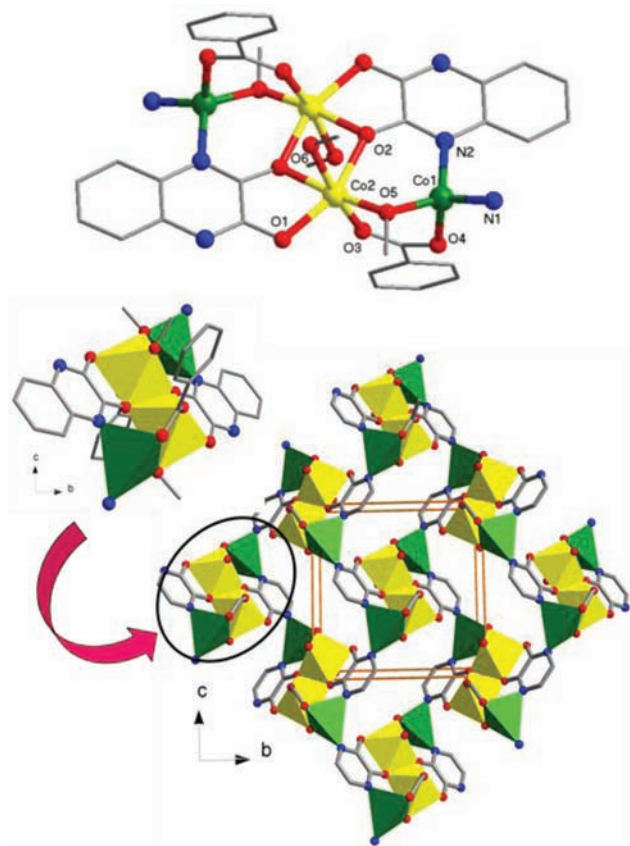


Figure 1. (Top) Crystal structure drawing of the Co₄ unit of **1**. Hydrogen atoms are omitted for clarity. (bottom) Polyhedral representation of the Co₄ cluster-based 2D layer of **1** viewed along the *a* axis.

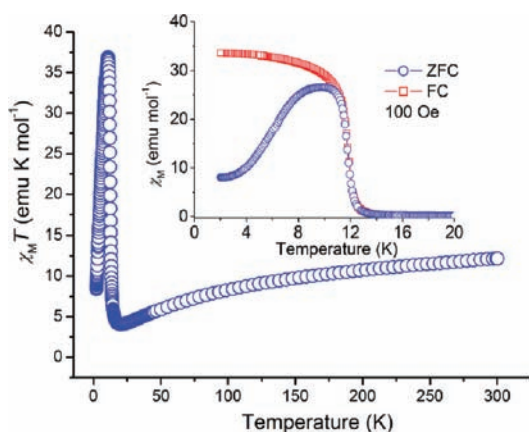


Figure 2. Plots of $\chi_M T$ vs *T* for **1** in an applied field of 1 kOe from 1.8 to 300 K and FC/ZFC magnetization for **1** at an applied field of 100 Oe (inset).

between the effective spins of adjacent Co^{II} ions. This conclusion was also reinforced by field dependence $\chi_M T$ vs *T* measurements (see Figure S5 in the Supporting Information). This magnetic behavior suggests that the Co₄ unit is present in a frustrated nonzero spin ground state and that each Co₄ unit is coupled antiferromagnetically, as the result of a noncanceled spin alignment, originating from spin-canted magnetism. The observed spin canting may be attributed to the magnetic anisotropy of Co^{II} ions and to an absence structural symmetry

at low temperature, as has been previously reported for some spin-canting MOFs.⁶

To further investigate the possible phase transition suggested by the increase in the $\chi_M T$ curve, zero-field-cooled (ZFC) and field-cooled (FC) magnetization were measured from 2.0 to 50 K under 100 Oe. As can be seen from the inset in Figure 2, the ZFC magnetization shows a broad peak at 10 K and diverges from FC magnetization at about 12 K. This type of divergence reveals a possible transition from a paramagnetic state to either a long-range-ordered, spin-glass, or superparamagnetic state. The hysteresis loops at 2.0, 3.0, 6.0, 8.0, and 10 K and isothermal magnetization *M*(*H*) at 2.0 K with a field of up to 70 kOe were measured, and the data are represented in Figures 3

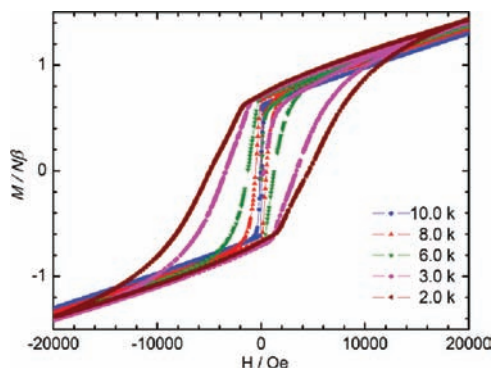


Figure 3. Magnetization vs field hysteresis loop measurements of **1** at the indicated temperatures at a scan rate of 0.03 T s⁻¹.

and S6 in the Supporting Information, respectively. A hysteresis loop can be clearly seen, even at 8 K. The coercivity fields and remnant magnetizations are $H_C = 4.77, 3.28, 2.43,$ and 1.79 kOe and $M_R = 0.523, 0.519, 0.515,$ and 0.513 Nβ (β = Bohr magneton) at 2.0, 3.0, 6.0, and 8.0 K, respectively. At 2.0 K, the magnetization shows a step at around 1300 Oe, which then increases slowly to 2.12 Nβ at 70 kOe, which is far from the saturated magnetization value $M_S = 13$ Nβ but close to a Co^{II} spin of 2.17 Nβ resulting from a Kramers $S = 1/2$ and average $g = 13/3$.⁷ The existence of a step in the magnetization at low field is indicative of metamagnetism in **1** (Figure S6 in the Supporting Information), and we attribute this behavior to the depression of the antiparallel configuration between spin-canting layers.^{3,8}

Furthermore, alternating-current (ac) susceptibilities under dc ($H_{dc} = 0$ Oe) and ac ($H_{ac} = 3.5$ Oe) fields were measured from 2 to 20 K with a frequency from 10 to 10 000 Hz. The results show strong peaks in both the in-phase (χ_M') and out-of-phase (χ_M'') parts. Upon decreasing temperature, the in-phase signals χ_M' behave similarly to the ZFC curves; i.e., they increase abruptly at around 12 K, reach a maximum at about 11.5 K, and then decrease slowly toward zero (Figure S7 in the Supporting Information). The out-of-phase signals χ_M'' increase steadily to a maximum of approximately 11.5–9.5 K (see Figure 4). The $d\chi_M'/dT$ curves all show peaks at $T_C = 11.6$ K, consistent with the dc measurements. While this behavior would not be expected to occur for true three-dimensionally ordered magnets, it is consistent with a spin-glass magnet. The peak temperatures of χ_M' can be measured by the parameter $\phi = (\Delta T_p/T_p)/\Delta(\log f) = 0.003$, indicating spin-glass behavior.¹⁰ Moreover, the specific heat was measured under a zero applied field of 1.8–20 K (see Figure S8 in the Supporting Information). A shoulder peak was observed at around 12 K,

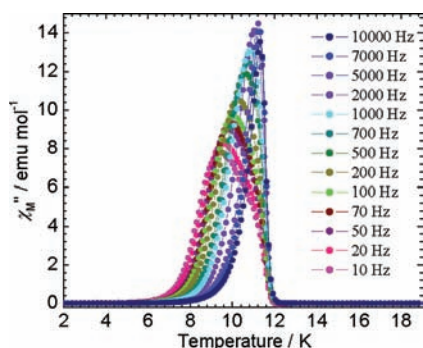


Figure 4. Plots of χ_M'' vs temperature for a powdered sample of complex **1** in a 3.5 G ac field. The data were collected in an ac field oscillating at the indicated frequency.

which is in good agreement with the occurrence of a spontaneous magnetization temperature, T_C , deduced from the ac susceptibility signal. This observation from specific heat measurements may be due to spin-glass freezing.¹¹ Moreover, the ac data were collected using the same sample that had been heated at 350 °C for 2 h (see Figure S9 in the Supporting Information), which shows the absence of out-of phase signals and changing in-phase signals, excluding the contribution of the behavior of cobalt(II) oxide.

In conclusion, we describe the preparation of a new cluster-based 2D layer with an unusual repeating unit of $[\text{Co}_4(\text{OMe})_2(\text{O}_2\text{CPh})_2(\text{dhq})_2(\text{MeOH})_2]_n$, prepared using a 2,3-dihydroxyquinoline ligand (H_2dhq), which exhibits spin-glass behavior resulting from spin-canting magnetic interactions. It thus appears that the H_2dhq ligand utilized here has the potential for generating novel frameworks with promising structural features and properties. Furthermore, these results provide strong justification for further studies of H_2dhq and derivatives thereof with other transition metals and could lead to the development of even more interesting magnetic MOFs.

■ ASSOCIATED CONTENT

■ Supporting Information

X-ray crystallographic data in CIF format, detailed experimental procedures, additional crystallographic diagrams, and a magnetic diagram. This material is available free of charge via the Internet at <http://pubs.acs.org>.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: lu@chem.sinica.edu.tw. Fax: (+886) 2-27831237.

■ ACKNOWLEDGMENTS

We thank Academia Sinica and the National Science Council of Taiwan for financial support and Prof. Hui-Lien Tsai at the National Cheng Kung University for PPMS support.

■ REFERENCES

- (1) (a) Öhrström, L.; Larsson, K. *Molecule-Based Materials: The Structural Network Approach*; Elsevier: Amsterdam, The Netherlands, 2005. (b) Kurmoo, M. *Chem. Soc. Rev.* **2009**, *38*, 1353.
- (2) (a) Manson, J. L.; Kmety, C. R.; Palacio, F.; Epstein, A. J.; Miller, J. S. *Chem. Mater.* **2001**, *13*, 1068. (b) Zheng, Y. Z.; Tong, M. L.; Xue, W.; Zhang, W. X.; Chen, X. M.; Grandjean, F.; Long, G. J. *Angew. Chem., Int. Ed.* **2007**, *46*, 6076. (c) Ferlay, S.; Mallah, T.; Ouahes, R.; Veillet, P.; Verdager, M. *Nature* **1995**, *378*, 701. (d) James, S. L. *Chem. Soc. Rev.* **2003**, *32*, 276. (e) Yoon, J. W.; Jhung, S. H.; Hwang, Y.

K.; Humphrey, S. M.; Wood, P. T.; Chang, J. S. *Adv. Mater.* **2007**, *19*, 1830. (f) Mukherjee, S.; Lan, Y.; Kostakis, G. E.; Clérac, R.; Anson, C. E.; Powell, A. K. *Cryst. Growth Des.* **2009**, *9*, 577.

(3) (a) Jensen, P.; Batten, S. R.; Fallon, G. D.; Moubaraki, B.; Murray, K. S.; Price, D. J. *Chem. Commun.* **1999**, 177. (b) Martin, S.; Barandika, M. G.; Lezama, L.; Pizarro, J. L.; Serna, Z. E.; Ruiz de Larramendi, J. I.; Arriortua, M. I.; Rojo, T.; Cortés, R. *Inorg. Chem.* **2001**, *40*, 4109. (c) Miyasaka, H.; Nakata, K.; Sugiura, K.; Yamashita, M.; Clérac, R. *Angew. Chem., Int. Ed.* **2004**, *43*, 707. (d) Miyasaka, H.; Nakata, K.; Lecren, L.; Coulon, C.; Nakazawa, Y.; Fujisaki, T.; Sugiura, K.; Yamashita, M.; Clérac, R. *J. Am. Chem. Soc.* **2006**, *128*, 3770.

(4) Satisfactory elemental analytical data found (calcd) for $\text{C}_{17}\text{H}_{16}\text{Co}_2\text{N}_2\text{O}_6$ (1): C, 44.14 (44.18); H, 3.60 (3.49); N, 6.28 (6.06). Crystal data for 1: $\text{C}_{17}\text{H}_{16}\text{Co}_2\text{N}_2\text{O}_6$, $M = 462.18$, monoclinic, space group $P2_1/c$, $a = 14.2780(7)$ Å, $b = 11.1546(6)$ Å, $c = 11.3505(6)$ Å, $\beta = 104.082(3)^\circ$, $V = 1753.42(16)$ Å³, $Z = 4$, $T = 150(2)$ K, $R (R_w) = 0.0433$ (0.1033).

(5) (a) Liu, Y.-H.; Tsai, H.-L.; Lu, Y.-L.; Wen, Y.-S.; Wang, J.-C.; Lu, K.-L. *Inorg. Chem.* **2001**, *40*, 6426. (b) Zeng, M.-H.; Wu, M.-C.; Liang, Y.-L.; Zhou, H.; Chen, X.-M.; Ng, S. W. *Inorg. Chem.* **2007**, *46*, 7241. (c) Zeng, M.-H.; Wang, B.; Wang, X. Y.; Zhang, W.-X.; Chen, X.-M.; Gao, S. *Inorg. Chem.* **2006**, *45*, 7069. (d) Zhou, Y.-L.; Meng, F.-Y.; Zhang, J.; Zeng, M.-H.; Liang, H. *Cryst. Growth Des.* **2009**, *9*, 1402.

(6) (a) Wang, X.-Y.; Sevov, S. C. *Inorg. Chem.* **2008**, *47*, 1037. (b) Zheng, Y.-Z.; Tong, M.-L.; Zhang, W.-X.; Chen, X.-M. *Angew. Chem., Int. Ed.* **2006**, *45*, 6310. (c) Mondal, K. C.; Kostakis, G. E.; Lan, Y.; Anson, C. E.; Powell, A. K. *Inorg. Chem.* **2009**, *48*, 9205. (d) Pali, A. V.; Reu, O. S.; Ostrovsky, S. M.; Klokishner, S. I.; Tsukerblat, B. S.; Sun, Z.-M.; Mao, J.-G.; Prosvirin, A. V.; Zhao, H.-H.; Dunbar, K. R. *J. Am. Chem. Soc.* **2008**, *130*, 14729. (e) Zeng, M.-H.; Zhang, W.-X.; Sun, X.-Z.; Chen, X.-M. *Angew. Chem., Int. Ed.* **2005**, *44*, 3079. (f) Gao, E.-Q.; Wang, Z.-M.; Yan, C.-H. *Chem. Commun.* **2003**, 1748. (g) Wang, X.-Y.; Wang, Z.-M.; Gao, S. *Inorg. Chem.* **2008**, *47*, 5720.

(7) de Jongh, L. J., Ed. *Magnetic Properties of Layered Transition Metal Compounds*; Kluwer Academic Publishers: Dordrecht, The Netherlands, 1990.

(8) (a) Pereira, C. L. M.; Pedrosa, E. F.; Stumpf, H. O.; Novak, M. A.; Ricard, L.; Ruiz-García, R.; Rivière, E.; Journaux, Y. *Angew. Chem., Int. Ed.* **2004**, *43*, 956. (b) Gao, E.-Q.; Liu, P.-P.; Wang, Y.-Q.; Yue, Q.; Wang, Q.-L. *Chem.—Eur. J.* **2009**, *15*, 1217.

(9) (a) Kahn, O. *Molecular Magnetism*; VCH: Weinheim, Germany, 1993. (b) Carlin, R. L.; van Duijneveldt, A. J. *Magnetic Properties of Transition Metal Compounds*; Springer: New York, 1977.

(10) (a) $\phi = 0.01$ is a typical value for a spin glass. For details, see: Mydosh, J. A. *Spin Glasses: An Experimental Introduction*; Taylor & Francis: London, 1993. (b) Clérac, R.; O'Kane, S.; Cowen, J.; Ouyang, X.; Heintz, R.; Zhao, H.; Bazile, M. J.; Dunbar, K. R. *Chem. Mater.* **2003**, *15*, 1840. (c) Huang, F. P.; Tian, J. L.; Li, D. D.; Chen, G. J.; Gu, W.; Yan, S. P.; Liu, X.; Liao, D. Z.; Cheng, P. *Inorg. Chem.* **2010**, *49*, 2525.

(11) Chernova, N. A.; Ma, M.; Xiao, J.; Whittingham, M. S.; Breger, J.; Grey, C. P. *Chem. Mater.* **2007**, *19*, 4682.